

LETTERS TO THE EDITOR

On the Hardness of a New Boron Phase, Orthorhombic γ -B₂₈

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Abstract—Measurements of the hardness of a new high-pressure boron phase, orthorhombic γ -B₂₈, are reported. According to the data obtained, γ -B₂₈ has the highest hardness (~ 50 GPa) of all known crystal-line modifications of boron.

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Among the 16 polymorphous modifications of boron mentioned in the literature [1] only three phases seem to correspond to the pure element. They are: the α -B₁₂ rhombohedral low-temperature [2], β -B₁₀₆ rhombohedral high-temperature low-pressure [3], and t-B₁₉₂ tetragonal high-temperature [4] phases. Recently a new high-pressure boron phase, namely, orthorhombic γ -B₂₈ [5] has been synthesized. The structure of this phase (Fig. 1a) has been established [5] by *ab initio* calculations in the framework of the USPEX evolutionary algorithm [6]. At present only the hardness of the α -B₁₂ ($H_V = 42$ GPa [7]) and β -B₁₀₆ ($H_V = 45$ GPa [8]) phases has been measured and the experimental values are in good agreement with the values (39.2 and 43.8 GPa, respectively) calculated by us in the framework of the thermodynamic models of hardness [9, 10]. In the present study the hardness of polycrystalline orthorhombic γ -B₂₈ is measured for the first time.

As the initial material for synthesis of γ -B₂₈, highly crystalline β -B₁₀₆ (99.995 at %) was used. The γ -B₂₈ phase was synthesized in a multianvil two-stage apparatus (a 1000-ton press Max Voggenreiter/Walker module) at pressures from 15 to 20 GPa and temperatures from 1800 to 2000 K for 10–30 min. To isolate the sample from the high-pressure cell elements, capsules of pyrolytic boron nitride, which does not react with crystalline boron at temperatures below 2000 K [11], have been used. According to X-ray diffractometry (TEXT 3000, INEL) and elemental analysis (SX-50 Camebax, Cameca), the prepared samples are defined as crystalline γ -B₂₈ (the orthorhombic system, space group *Pnmm*, $a = 5.054$ Å, $b = 5.612$ Å, $c = 6.966$ Å) that does not contain impurities.

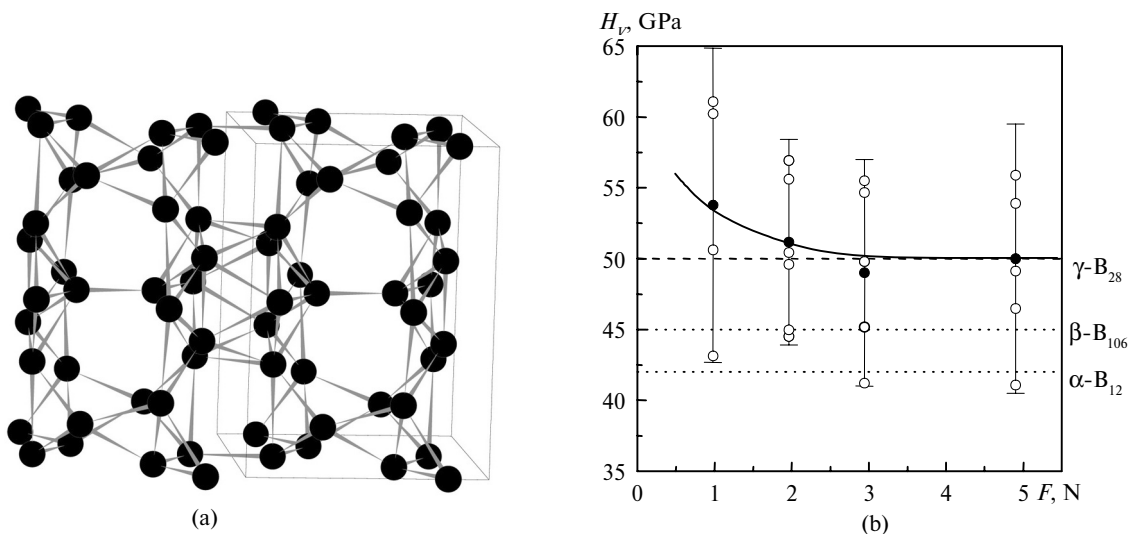


Fig. 1. The γ -B₂₈ phase crystal structure (a) and the phase hardness vs. load (b). The centres of gravity of B₁₂ icosahedra and B₂ dumbbells in the structure of the new phase form a NaCl-type structure [5].

The Vickers hardness was measured using a Struers Duramin-20 microhardness tester under loads (F) from 1 to 20 N and at an indentation time of 20 s. At least 4 indents were spaced at about 200 μm intervals. At $F > 10$ N the appearance of long cracks and splallings obstructed accurate estimation of the hardness.

According to the data obtained (Fig. 1b), polycrystalline phase has a hardness of 50 (11) GPa, which is markedly higher than the hardness of other boron modifications [7, 8] and agrees well with the value of 48.8 GPa that was calculated in the framework of the thermodynamic model of hardness [9, 10]. Thus, new high-pressure phase $\gamma\text{-B}_{28}$ has the highest hardness among the known boron crystalline modifications, which stems from its highest density (2.544 g/cm^3).

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REFERENCES

1. Douglas, B.E. and Ho, S.-M., Structure and Chemistry of Crystalline Solids, N.Y.: Springer, 2006.
2. Decker, B. F. and Kaspe, J. S., The Crystal Structure of a Simple Orthorhombic Form Boron, *Acta Cryst.*, 1959, vol. 12, pp. 503–506.
3. Hoard, J.L., Sullenger, D.B., Kennard, C.H. L., and Hughes, R.E., The Structure of Beta-Orthorhombic Boron, *J. Solid State Chem.*, 1970, vol. 1, no. 2, pp. 268–277.
4. Vlasse, M., Naslain, R., Kasper, J.S., and Ploog, K., Crystal Structure of Tetragonal Boron Related to $\alpha\text{-AlB}_{12}$, *J. Solid State Chem.*, 1979, vol. 28, no. 3, pp. 289–301.
5. Oganov, A.R., Chen J., Gatti C., et al., Ionic High-Pressure Form of Elemental Boron, *Nature*, 2008, manuscript 2007-01-00879, in press, pp. 1–4.
6. Oganov, A.R. and Glass, C.W., Crystal Structure Prediction Using Ab Initio Evolutionary Algorithms: Principles and Applications, *J. Chem. Phys.*, 2006, vol. 124, pp. 244704.
7. Amberger, E. and Stumpf, W., Boron, *Gmelin Handbook of Inorganic Chemistry*, Berlin: Springer-Verlag, 1981, pp. 112–238.
8. Bolmgren, H., Lundström, T., and Okada, S., Structure Refinement of the Boron Suboxide B_6O by the Rietveld Method, *Boron-Rich Solids*, Proc. AIP Conference, New York: American Institute of Physics, 1991, vol. 231, pp. 197–200.
9. Mukhanov, V.A., Kurakevych, O.O., and Solozhenko, V.L., Thermodynamic Aspects of Materials' Hardness: Prediction of Novel Superhard High-Pressure Phases, *High Press. Res.*, 2008, DOI: 10.1080/08957950802429052 (manuscript GHPR343073), pp. 1–7.
10. Mukhanov, V.A., Kurakevych, O.O., and Solozhenko, V.L., The Interrelation between the Hardness and Compressibility of Substances and their Structure and Thermodynamic Properties, *Sverkhtverdye Materialy*, 2008, no. 6, pp. 10–22 [*J. Superhard Materials*, 2008, vol. 30, no. 6, pp. 368–378].
11. Solozhenko, V.L., Le Godec, Y. and Kurakevych, O.O., Solid-State Synthesis of Boron Subnitride, B_6N : Myth or Reality?, *C. R. Chimie*, 2006, vol. 9, nos. 11–12, pp. 1472–1475.